

Investigations into nicotinic acetylcholine receptor (nAChR) antagonists: synthesis of a sub-unit of methyllycaconitine

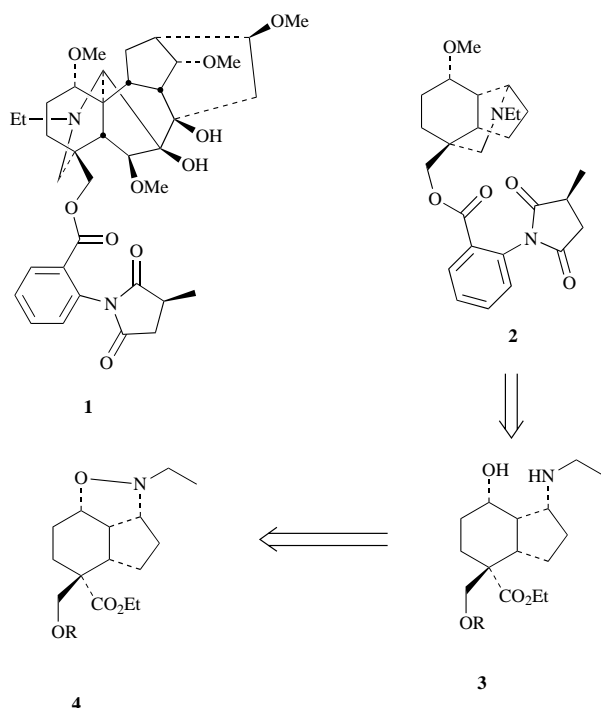
Lynn C. Baillie,^a John R. Bearder,^b John A. Sherringham^a and Donald A. Whiting^{*,a}

^a Department of Chemistry, Nottingham University, Nottingham, UK NG7 2RD

^b Shell Thornton Research Centre, P.O. Box 1, Chester, UK CH1 3SH

A potentially toxophoric subunit of methyllycaconitine has been synthesised from penta-1,4-dien-3-ol in 14 steps and 5% overall yield.

Methyllycaconitine (MLA) **1**, a C₁₉ diterpenoid alkaloid, is found in the *Delphinium* and *Aconitum* species,^{1,2} and has proved highly toxic to mammals and insects. In both cases, it acts on nicotinic acetylcholine receptors (nAChR) inhibiting neurotransmission and inducing paralysis³ and it is reported to be the most potent non-protein competitive nAChR antagonist discovered to date.⁴ MLA has recently acquired extensive use in distinguishing nAChR subtypes, which may be implicated in Alzheimer's disease,⁵ and has been found to have high affinity for the $\alpha 7$ subtype;⁶ it is more selective than α -bungarotoxin,⁷ a polypeptide neurotoxin isolated from snake venom, but has comparable potency.

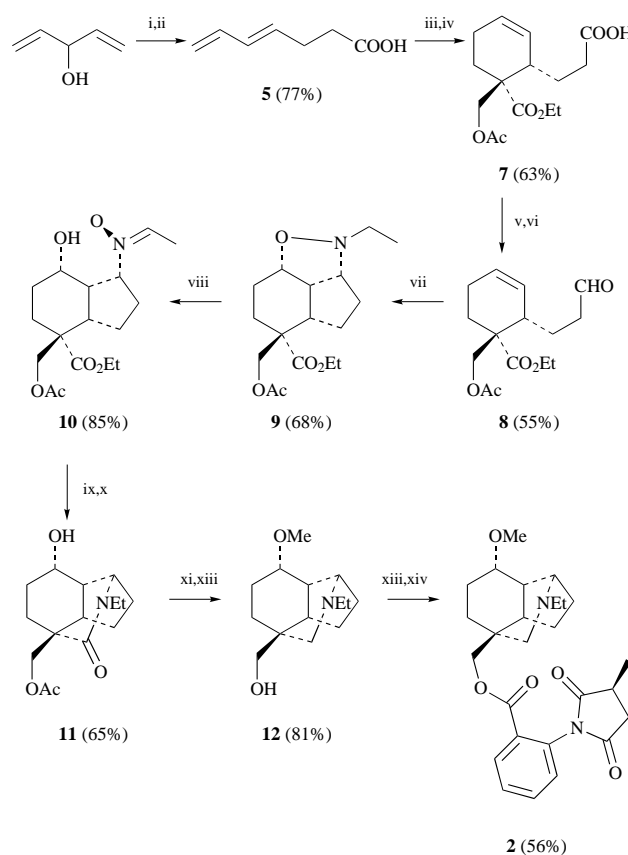


Scheme 1 Retrosynthesis

The high toxicity to animals of MLA disbars it as an agrochemical; however, if the inhibitory action is localised in a toxophoric section, a smaller subunit could find practical application if of significantly lower toxicity to mammals. To this end a number of methyllycaconitine analogues have been investigated. Structure-activity relationship investigations have shown that the methyl group on the succinimido ring⁸ and the ethyl

group of the tertiary amine⁹ are important and this led us to investigate the synthesis of the A/E/F tricyclic fragment **2**. Previous work in this laboratory¹⁰ led us to use a strategy based on the disconnection of the C–N bond to the *cis*-fused 6,5-bicyclic **3**, which could in turn be derived from the isoxazolidine **4** (Scheme 1).

Scheme 2 shows the forward synthesis. Penta-1,4-dien-3-ol



Scheme 2 Reagents and conditions: i, MeC(OEt)₃, EtCO₂H, 142 °C, 3 h; ii, KOH, MeOH; iii CH₂C(CH₂OH)CO₂Et **6**, LiCl, aq NaHCO₃, 60 °C, 120 h; iv Ac₂O, pyr, 12 h; v, (a) (Pr^t)₂NEt, Bu^tOCOCI, DME; (b) NaBH₄; vi TPAP, NMO; vii, EtNH₂·CF₃CO₂H, PhH, reflux, 3 h; viii, MCPBA, CH₂Cl₂, 0 °C; ix, PtO₂, H₂, 1 atm, 48 h; x, xylene, reflux, 24 h; xi, MeI, Ag₂O, reflux, 24 h; xii, LiAlH₄, dioxane, reflux, 12 h; xiii, isatoic anhydride, DMAP, DMF; xiv, (S)-(-)-methyl succinic anhydride, xylene

was heated with triethyl orthoacetate and a catalytic amount of propionic acid and the resulting ester was subjected to alkaline hydrolysis to afford the (*E*)-hepta-4,6-dienoic acid **5**.¹¹ Diels-Alder reaction of the sodium salt of the acid with 2 equivalents of the methacrylate **6**¹² in aq. 5 M lithium chloride¹³ yielded the

cyclohexene acid as 3.3:1 mixture of stereoisomers (*endo*:*exo*).[†] Much experimentation was required to discover the most suitable substituents, dieneophile, and reaction conditions to minimise polymerisation. Acetylation followed by separation of the diastereomers gave the desired *endo* acid **7**. The acid was converted into the corresponding alcohol in a one-pot procedure whereby the acid was first converted to the mixed anhydride with diisopropylethylamine and isobutyl chloroformate and then reduced with sodium borohydride.¹⁴ Oxidation of the alcohol with TPAP¹⁵ gave the aldehyde **8**, which was heated to reflux with *N*-ethylhydroxylamine in benzene to provide the isoxazolidine **9**. All attempts to cleave the N–O bond reductively were, to our surprise, unsuccessful. Thus we deployed a two-step procedure in which the isoxazolidine was oxidised to the *N*-oxide with MCPBA; a spontaneous elimination/ring-opening resulted in formation of the nitrone **10**. This was reduced to the amine by catalytic hydrogenation over platinum. Subsequent reactions showed that the stereochemical integrity of the C–N bond was retained throughout this sequence. Ring-closure to the amide **11** was effected by heating to reflux in xylene for 24 h. *O*-Methylation of the alcohol functionality was achieved by refluxing with methyl iodide in the presence of silver(I) oxide: performing the *O*-methylation step before the reduction of the amide circumvented problems which arose in the model system in which the quaternary amine salt was formed as a significant side product.¹⁰ Reduction of the amide to the amine **12**, with concomitant removal of the acetyl protecting group was effected with LiAlH₄. Reaction of the alcohol functionality with isatoic anhydride gave the anthranilate. The succinimido ring was incorporated by refluxing this intermediate with (*S*)-methyl succinic anhydride in xylene to give the desired A/E/F tricyclic fragment **2** as a mixture of diastereomers.[‡]

The stereochemistry generated by the Diels–Alder reaction and the 1,3-dipolar addition was confirmed by X-ray crystallo-

graphic studies on the nitrone. In addition the heterocyclic ring closure to the amide **11** would have not been possible if the relative stereochemistry of the amine and the ester functionalities had been incorrect.

Thus starting from the commercially available penta-1,4-dien-3-ol, we have synthesised the desired A/E/F tricycle **2** in 14 steps and 5% overall yield. This ester models the A/E/F tricyclic system of methyllycaconitine and contains 6 stereogenic centres. Biological testing of this compound is currently underway.

Acknowledgements

We thank Dr W.-S. Li for X-ray crystallographic analysis, the EPSRC and Shell Research Ltd. for a CASE award to LCB and the University of Nottingham for a Fellowship to JAS.

References

- 1 M. S. Yusunov, *Nat. Prod. Rep.*, 1991, **8**, 499.
- 2 S. W. Pelletier, N. V. Mody, K. Varughese, J. A. Maddry and H. K. Desai, *J. Am. Chem. Soc.*, 1981, **103**, 6536.
- 3 A. Drasdo, M. Caulfield, D. Bertrand, S. Bertrand and S. Wonnacott, *Mol. Cell. Neurosci.*, 1992, **3**, 237.
- 4 S. Wonnacott, E. X. Albuquerque and D. Bertrand, *Methods in Neuroscience*, 1993, **12**, 263.
- 5 S. P. Arneric, J. P. Sullivan, M. W. Decker, J. D. Brioni, A. W. Bannon, C. A. Briggs, D. Donnelly-Roberts, R. J. Radek, K. C. Marsh, J. Kyncl, M. Williams and J. J. Buccafusco, *Alzheimer Dis. Assoc. Disord.*, 1995, **9**, Suppl. 2, 50.
- 6 E. Palma, S. Bertrand, T. Binzoni and D. Bertrand, *J. Physiol.*, 1996, **491**, 151.
- 7 (a) P. B. S. Clarke, *Trends Pharmacol. Sci.*, 1992, **13**, 407; (b) S. P. H. Alexander and J. A. Peters, *Trends Pharmacol. Sci., Receptor and Ion Channel Nomenclature Supplement*, 1997, 5.
- 8 J. M. Jacyno, J. S. Harwood, N. H. Lin, J. E. Campbell, J. P. Sullivan and M. W. Holland, *J. Nat. Prod.*, 1996, **59**, 709.
- 9 G. D. Manners, K. E. Panter and S. W. Pelletier, *J. Nat. Prod.*, 1995, **58**, 863.
- 10 L. C. Baillie, J. R. Bearder and D. A. Whiting, *J. Chem. Soc., Chem. Commun.*, 1994, 2487.
- 11 W. M. Roush, H. R. Gillis and A. I. Ko, *J. Am. Chem. Soc.*, 1982, **104**, 2269.
- 12 J. Villieras and M. Rambaud, *Synthesis*, 1982, 924.
- 13 R. Breslow and U. Maitra, *Tetrahedron Lett.*, 1984, **25**, 1239.
- 14 M. Rodriguez, M. Llinares, S. Doulit, A. Heitz and J. Martinez, *Tetrahedron Lett.*, 1991, **32**, 923.
- 15 S. V. Ley, J. Norman, W. P. Griffith and S. P. Marsden, *Synthesis*, 1994, 639.

Paper 7/04447J

Received 24th June 1997

Accepted 17th July 1997

[†] To a solution of (*E*)-hepta-4,6-dienoic acid **5** (9.86 g, 78 mmol) in water (30 ml) at 25 °C was added sodium hydrogen carbonate (6.57 g, 78 mmol) portionwise over 30 min. After gas evolution had subsided, ethyl 2-(hydroxymethyl)acrylate **6** (20.34 g, 156 mmol) and lithium chloride (6.14 g, 145 mmol) were added. The mixture was warmed to 60 °C and stirred for 120 h. The mixture was extracted with diethyl ether to remove unreacted methacrylate, then acidified to pH3 with 2 M aqueous hydrochloric acid and extracted into diethyl ether. The extracts were washed with brine, dried and concentrated. The product was subjected to column chromatography (light petrol–ethyl acetate, 7:3) to yield the desired compound as a colourless oil (18.16 g, 91%) (*endo*:*exo* 3.3:1, as determined by NMR spectroscopy).

[‡] All new compounds gave satisfactory spectroscopic and analytical data.